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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q_1 and Q_2 separated by a distance r is given by $Q_1 Q_2 / (4\pi\epsilon_0 r^2)$ First we consider the interaction between Na^+ ion and $6Cl^-$ ions at distance r Applying Coulomb's law we have $Q_1 Q_2 = (-6e)(+e) = -6e^2$ $PE_1 = = = 4\pi\epsilon_0 r^2$ $4\pi\epsilon_0 r^2$ $4\pi\epsilon_0 r^2$ Similarly, we now consider 12 Na^+ ions as second nearest neighbors at a distance r $Q_1 Q_2 = (+12e)(+e) = 12e^2 = PE_2 = = 4\pi\epsilon_0 r^2$ $4\pi\epsilon_0 r^2$ $4\pi\epsilon_0 r^2$ and Cl^- ...

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Ernest Rutherford (1871-1937) 1.37 fSolutions to Principles of Electronic Materials and Devices: 2nd Edition (Summer 2001) Chapter 2 Second Edition (2001 McGraw-Hill) Chapter 2.1 Electrical conduction Na is a monovalent metal (BCC) with a density of 0.9712 g cm^{-3} . Its atomic mass is 22.99 g mol^{-1} .

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